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## Safety Data Sheet

### 1 IDENTIFICATION

#### Product identifier

**Trade name:** Silver Brazing Alloys

**Product size:** Bare & Flux-Coated Wire or Rods

**Synonyms:** Cadmium free, 2%, 5%, 15%, 45%, 56% (Bare Silver), 45%, 56% (Flux Coated).

#### Other means of identification

SDS Number: 02-1

#### Recommended use and restriction on use

**Recommended use:** Metal Brazing

#### Manufacturer/Importer/Supplier/Distributor information

##### Importer:

NEW ZEALAND

Harris Products Group

47 Edmundson St, Onekawa, Napier

New Zealand 4110

(06) 83405875

**Safety Data Sheet Questions:** [sales@harrisnz.com](mailto:sales@harrisnz.com)

**Website:** <http://www.harrisproductsgroup.co.nz>

**New Zealand National Poisons Centre/Helpline (24 hours) 0800 POISON (0800 764 766)**

**Fire Service - Ambulance – 111**

##### AUSTRALIA

Harris Products Group

14 Queensland Rd

Darra, QLD, Australia 4076

(07) 33753670

**Safety Data Sheet Questions:** [sales@hgea.com.au](mailto:sales@hgea.com.au)

**Website:** <http://www.harrisproductsgroup.com.au>

### 2 HAZARD(S) IDENTIFICATION

#### GHS classification of the substance/mixture.

Classified according to the Globally Harmonised System of Classification and labelling of Chemicals (GHS) including Work, Health and Safety regulations, Australia.

**HAZARDOUS CHEMICAL. NON-DANGEROUSE GOODS. According to the WHS Regulations and the ADG Code.**

#### HAZAD RATINGS

0 = Minimum

1 = Low

2 = Moderate

3 = High

4 = Extreme

Flammability	0
Toxicity	2
Body Contact	3
Reactivity	0
Chronic	

Label Elements  
GHS label elements



Skin Sensitiser



Organ Toxicity



Aquatic Hazard

**Signal Word**                      **WARNING**

**Precautionary Statement**

**H317**    May cause an allergic skin reaction  
**H351**    Suspected of causing cancer  
**H373**    May damage organs  
**H410**    Very toxic to aquatic life

**Prevention**

**P201**    Obtain special instructions before use.  
**P260**    Do not breath in dust, gas, mist, fumes, vapours, sprays  
**P280**    Wear face protection, protective gloves and clothing, eye protection.  
**P281**    Use protective equipment as needed.

**Response**

**P308**    If concerned about exposure: Get medical advice  
**P363**    Wash contaminated protective clothing before using again  
**P302**    If contact with skin: wash with soap and water  
**P314**    If feel unwell: get medical advice

**Storage**

**P405**    Lock up storage

**Disposal**

**P501**    Dispose in accordance with local authority regulations

**Additional information:**

**Other hazards which do not result in GHS classification:**

Heat rays (infrared radiation) from flame or hot metal can injure eyes. Overexposure to brazing fumes and gases can be hazardous. Read and understand the manufacturer's instructions, Safety Data Sheets and the precautionary labels before using this product.

### 3 Composition/information on ingredients

**Chemical characterization: Mixtures**

**Description:** Mixture: consisting of the following components.

<b>Dangerous components:</b>		
<b>CAS</b>	<b>Name</b>	<b>Proportion</b>
7440-50-8	Alloy comprising copper	10-60%
7440-2204	silver	10-60%
7440-31-5	tin	0-20%
7440-66-6	zinc	10-60%
7440-02-0	nickel	<10%
7440-21-3	silicon	0-1%
Flux Coating on Rods		
10043-3-3	Boric Acid	Not Established
7789-75-5	Proprietary fluoride compound	Not Established
Proprietary	Methacrylate/aliphatic compound	Not Established
7732-18-5	Water	Not Established
In Use Generate Welding Fumes		
7440-50-8	Copper fume	Not Available
1314-13-2	Zinc oxide fume	Not Available
7440-02-0	Nickel fume	Not Available

**Composition comments:**

The term “Dangerous Components” should be interpreted as a term defined in Hazard Communication standards and does not necessarily imply the existence of a hazard. The product may contain additional nonhazardous ingredients or may form additional compounds under the condition of use. Refer to Sections 2 and 8 for more information.

## 4 First Aid Measures

### Skin Exposure

If dusts or particulates generated by the flux coating on the flux coated products or fumes generated by brazing operations involving these products contaminate the skin, begin decontamination with running water. If molten material contaminates the skin, immediately begin decontamination with cold, running water. Minimum flushing is for 15 minutes. Victim must seek medical attention if any adverse reaction occurs.

### Eye Exposure

If dusts or particulates generated by the flux coating on the flux coated products or fumes generated by brazing operations involving these products enter the eyes, open victim's eyes while under gently running water. Use sufficient force to open eyelids. Have victim "roll" eyes. Minimum flushing is for 15 minutes. Victim must seek immediate medical attention.

### Inhalation

If dusts or particulates generated by the flux coating on the flux coated products or fumes generated by brazing operations involving these products are inhaled, remove victim to fresh air. If necessary, use artificial respiration to support vital functions.

### Ingestion

**If swallowed do not induce vomiting and call physician immediately!** Do not induce vomiting unless directed by medical personnel. Rinse mouth with water if person is conscious. Never give fluids or induce vomiting if person is unconscious, having convulsions, or not breathing.

### Medical conditions aggravated from exposure

Skin, respiratory, and kidney disorders may be aggravated by prolonged over-exposures to the dusts or fumes generated by these products.

### Recommendations to physicians

Treat symptoms and eliminate overexposure. Call for medical aid. Employ first aid techniques recommended by the Australian Red Cross

## 5 Fire-fighting measures

### Extinguishing media

#### Suitable extinguishing equipment:

There is no restriction on the type of extinguisher that may be used.  
Use extinguishing equipment suitable for surrounding area

### Special hazards arising from the product

Non-combustible. Not considered a risk, however containers may burn.  
Slight hazard when exposed to heat, flame and oxidisers.  
Decomposition can produce toxic fumes of nitrogen and metal oxides and silicon dioxide.

### Special protective equipment and precautions for fire fighters

Wear breathing apparatus plus protective gloves in event of a fire.  
Use fire fighting procedures suitable for surrounding area.

Welding arc and sparks can ignite combustibles and flammables.

### Additional information

Read and understand the Work Safe Australia Code of Practice on Welding Processes and “Standard for Fire Prevention During Welding, Cutting and Other Hot Work” before using this product. Section 274 of the Work Health and Safety Act (the WHS Act.)

## 6 Accidental release measures

### Personal precautions, protective equipment and emergency procedures

Personal Protective Equipment advice is contained in Section 8

### Environmental precautions

See section 12

Prevent waste from contaminating surrounding environment. Discard any product residue, disposable container or liner in environmentally acceptable manner, in full compliance with federal, state, and local regulations.

### Methods and materials for containment and cleaning up.

Clean up spills immediately

Wear protective clothing, safety eyewear, dust mask and protective gloves.

Secure load if safe to do so and bundle recoverable product.

## 7 Handling and storage

### Handling:

#### Precautions for safe handling

Avoid breathing dust and personal contact.

Ensure exhaust system at the workplace or well ventilated area.

Wear protective clothing and equipment – Section 8

#### Conditions for safe storage, including any incompatibilities

Store in original containers if possible

Store in a cool, dry, well ventilated area

Use suitable containers – manufacturers packaging, polyethylene container.

Make sure all containers are labelled and leak proof.

Do not store with acids as can produce flammable, explosive hydrogen gas. Also incompatible with the addition of concentrated nitric acid, acid chlorides, acid anhydrides and chloroformates.

## 8 Exposure controls/personal protection

### Exposure controls

Source	Material	TWA mg/m <sup>3</sup>	STEL mg/m <sup>3</sup>
Australia Exposure Standards	Copper (dust, mists Cu)	1	
Australia Exposure Standards	Copper (fume)	0.2	
Australia Exposure Standards	Copper (inspirable dust) not otherwise classified	10	
Australia Exposure Standards	Silver (Silver metal)	0.01	
Australia Exposure Standards	Tin (tin, metal)	2	
Australia Exposure Standards	Silicon (silicon (a))	10	

### Exposure Controls/Personal Protection

Source	Material	TWA mg/m <sup>3</sup>	STEL mg/m <sup>3</sup>
Australia Exposure Standards	Welding fumes (not otherwise classified)	5	
Australia Exposure Standards	Zinc oxide fume	5	10

### Engineering Controls

#### Copper Dusts and Mists:

**50 mg/m<sup>3</sup> or less** – A high efficiency particulate filter respirator with a full face piece or an supplied air respirator.

**2000 mg/m<sup>3</sup> or less** – A type C supplied air respirator with full face piece operated in pressure demand mode.

**Greater than 2000 mg/m<sup>3</sup>** or entry and escape from unknown concentrations – Self contained breathing apparatus with full face piece operated in pressure demand or other positive pressure mode.

**Silver Exposure-** Ventilation is recommended for controlling dust exposures including for silver. Appropriate administrative controls should also be used to ensure adequate maintenance for the ventilation system. Housekeeping policies should also be developed and used for further dust suppression and surface decontamination. With regard to silver metal overexposures between 0.01 mg/ operators must use engineering controls to reduce exposure to the permissible limit. As silver can cause eye and skin irritation, it is recommended that workers also where appropriate eye and skin protection compatible with the respirator.

**0.25mg/m<sup>3</sup>** – Supplied air operated respirator in continuous flow mode.

**0.50mg/m<sup>3</sup>** – Air purifying full face piece with tight fitting face piece operated in continuous flow mode.

**10mg/m<sup>3</sup>** – Supplied air operated in pressure demand full fitting face piece.

#### **Tin**

**2mg/m<sup>3</sup>** - Whenever possible the use of local exhaust ventilation or other engineering controls is the preferred method of controlling exposure to airborne dust and fume to meet established occupational exposure limits. Use good housekeeping and sanitation practices. Do not use tobacco or food in work area. Wash thoroughly before eating or smoking. Do not blow dust off clothing or skin with compressed air.

#### **Silicon**

5 mg/m<sup>3</sup> - Silicon dust appears to have little adverse effect on the lungs and is not implicated in the genesis of organic disease or in the production of toxic effects.

For aluminium oxide and pyrophoric grades of aluminium: Twenty-seven-year experience with aluminium oxide dust (particle size 96% 1,2 µm) without adverse effects either systemically or on the lung, and at a calculated concentration equivalent to 2 mg/m<sup>3</sup> over an 8-hour shift has led to the current recommendation of the TLV-TWA. The limit should also apply to aluminium pyro powders whose toxicity is reportedly greater than aluminium dusts and should be protective against lung changes. For aluminium oxide: The experimental and clinical data indicate that aluminium oxide acts as an "inert" material when inhaled and seems to have little effect on the lungs nor does it produce significant organic disease or toxic effects when exposures are kept under reasonable control. The concentration of dust, for application of respirable dust limits, is to be determined from the fraction that penetrates a separator whose size collection efficiency is described by a cumulative log-normal function with a median aerodynamic diameter of 4.0 µm (+-) 0.3 µm and with a geometric standard deviation of 1.5 µm (+-) 0.1 µm, i.e. Generally, less than 5 µm.

**Zinc Oxide Fumes** - Zinc oxide intoxication is characterised by general depression, shivering, headache, thirst, colic and diarrhoea. Exposure to the fume may produce metal fume fever characterised by chills, muscular pain, nausea and vomiting.

**Welding Fumes** - Local exhaust ventilation systems should be designed to provide a minimum capture velocity at the fume source of 0.5m/second away from the welder. Inlets and outlets should be kept clear at all times. Air from a local exhaust ventilation system should not be re-circulated into the workroom. This air should be discharged into the outside air away from other work areas and away from air conditioning inlets or compressors supplying breathing air.

Examples of local exhaust ventilation suitable for welding operations include:

Fixed installations, such as side-draught or down-draught tables and benches, and partially or completely enclosed booths. Portable installations, such as movable hoods that are attached to edible ducts.

Low volume high velocity fume extractors attached directly to the welding gun.

**Respiratory Protective Devices** - Respirators should be fitted for each person individually and if one is to be used by another operator, it must be disinfected and re fitted before use. The tightness of all connections and the condition of the face piece, headbands and valves should be checked before each use. Air supplied respirators may be required in some situations, e.g. confined spaces. For further information, refer to: AS/NZS 1716: *Respiratory protective devices* and be selected in accordance with AS/NZS 1715: *Selection, use and maintenance of respiratory protective equipment*.

#### **Personal protective equipment:**

##### **Respiratory Measures**

Use respirable fume respiratory or air supplied respirator when in confined space or local exhaust does not keep exposure below recommended exposure limit

### Ventilation

Use enough ventilation, local exhaust at the the flame or heat source, or both to keep the fumes and gases from the worker's breathing zone and the general area. Train the operator to keep his head out of the fumes. Keep exposure as low as possible.

### Breathing equipment:

Keep your head out of fumes. Use enough ventilation and local exhaust to keep fumes and gases from your breathing zone and the general area. An approved respirator should be used unless exposure assessments are below applicable exposure limits. Particulate mask should filter at least 99% of airborne particles.

### Protection of hands:



Thermally-protective gloves.

Suitable gloves can be recommended by the glove supplier.

The glove material has to be impermeable and resistant to the product/ the substance/ the preparation.

### Eye protection:



Wear helmet, glasses or face shield with filter lens protective screens, flash goggles to shield others. Start with shade too dark then go to lighter shade which gives sufficient view of weld zone.

### Body protection:



Protective work clothing

**Limitation and supervision of exposure into the environment** No special requirements.

**Risk management measures:** For maximum safety: Be certified for and wear a respirator at all times when welding and brazing.

## 9 Physical and chemical properties

### Information on basic physical and chemical properties

#### General Information

#### Appearance:

**Form:** Rod, strip, sheet, press parts, wire or rings

Does not mix with water

<b>Odour</b>	Not Available	<b>Viscosity</b>	Not Applicable
<b>Odour Threshold</b>	Not Available	<b>Molecular weight (g/mol)</b>	Not Applicable
<b>pH</b>	Not Applicable	<b>Taste</b>	Not Available
<b>Melting Range (C°)</b>	618 - 830	<b>Vapour Pressure (kPa)</b>	Not Available
<b>Boiling point &amp; boiling range</b>	Not Applicable	<b>pH as a solution</b>	Not Applicable
<b>Flash point</b>	Not Applicable	<b>Gas group</b>	Not Available
<b>Evaporation rate</b>	Not Available	<b>Explosive properties</b>	Not Available
<b>Flammability (solid, gas)</b>	Not Available	<b>Oxidising properties</b>	Not Available

<b>Upper/lower flammability or explosive limit</b>	Not Applicable	<b>Surface tension</b>	Not Applicable
<b>Vapour pressure</b>	Not Applicable		
<b>Vapour density</b>	Not Applicable		
<b>Relative density</b>	Not Available		
<b>Solubility</b>	Immiscible		
<b>Partition coefficient: n-octano/water</b>	Not Available		
<b>Auto-ignition temperature</b>	Not Applicable		
<b>Decomposition temperature</b>	Not Available		

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## 10 Stability and reactivity

**Reactivity:** Do not store with acids as can produce flammable, explosive hydrogen gas.

**Chemical stability:** Product is considered stable and hazardous polymerisation will not occur.

**Conditions to avoid:** Store in original containers if possible

Store in a cool, dry, well ventilated area

Use suitable containers – manufacturers packaging, polyethylene container.

Make sure all containers are labelled and leak proof.

**Incompatible materials:** Do not store with acids as can produce flammable, explosive hydrogen gas.

Also incompatible with the addition of concentrated nitric acid, acid chlorides, acid anhydrides and chloroformates.

**Hazardous decomposition products:**

Decomposition can produce toxic fumes of nitrogen and metal oxides and silicon dioxide.

Brazing fumes and gases cannot be classified simply. The composition and products: quantity of both are dependent upon the metal being joined, the process, procedure and filler metals and flux used. Other conditions which also influence the composition and quantity of the fumes and gases to which workers may be exposed include: coatings on the metal being joined (such as paint, plating, or galvanizing), the number of operators and the volume of the worker area, the quality and amount of ventilation, the position of the operator's head with respect to the fume and fumes from chemical fluxes used in some brazing operations.

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## 11 Toxicological information

### Acute Toxicity

#### Ingestion:

Unlikely route of exposure.

Soluble zinc salts produce irritation and corrosion of the alimentary tract (in a manner similar to copper salts) with pain, vomiting, etc. Delayed deaths have been ascribed to inanition (weakness and extreme weight loss resulting from prolonged and severe food insufficiency) following severe strictures of the oesophagus, and pylorus. Vomiting, abdominal cramps, and diarrhoea, in several cases with blood, have been observed after ingestion of zinc sulphate. Several cases of gastrointestinal disturbances have been reported after ingestion of zinc sulphate. A significant reduction in erythrocyte superoxide dismutase activity (47% decrease), haematocrit, and serum ferritin, compared to pre-treatment levels, occurred in female subjects who received supplements (as capsules) of 50 mg zinc/day as zinc gluconate for 10 weeks.

Numerous cases of a single oral exposure to high levels of copper have been reported. Consumption of copper-contaminated drinking water has been associated with mainly gastrointestinal symptoms including nausea, abdominal pain, vomiting and diarrhoea. A metallic taste, nausea, vomiting and epigastric burning often occur after ingestion of copper and its derivatives. The vomitus is usually green/blue and discolours contaminated skin.

Accidental ingestion of the material may be damaging to the health of the individual.

#### Inhalation:

Inhalation of fume may aggravate a pre-existing respiratory condition such as asthma, bronchitis, emphysema.

Copper poisoning following exposure to copper dusts and fume may result in headache, cold sweat and weak pulse. Capillary, kidney, liver and brain damage are the longer term manifestations of such poisoning. Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.

There is some evidence to suggest that the material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.

Inhalation hazard is increased at higher temperatures.

Inhalation of freshly formed zinc oxide particles sized below 1.5 microns and generally between 0.02 to 0.05 microns may result in "metal fume fever", with symptoms resembling influenza. Symptoms may be delayed for up to 12 hours and begin with the sudden onset of thirst, and a sweet, metallic or foul taste in the mouth.

Regular exposure to nickel fume, as the oxide, may result in "metal fume fever" a sometimes debilitating upper respiratory tract condition resembling influenza.

Symptoms include malaise, fever, weakness, nausea and may appear quickly if operations occur in closed or poorly ventilated areas.

Inhalation of freshly formed metal oxide particles sized below 1.5 microns and generally between 0.02 to 0.05 microns may result in "metal fume fever".

**Skin Contact:** Exposure to copper, by skin, has come from its use in pigments, ointments, ornaments, jewellery, dental amalgams and IUDs (intra-uterine devices), and in killing fungi and algae. Although copper is used in the treatment of water in swimming pools and reservoirs, there are no reports of toxicity from these applications. Reports of allergic contact dermatitis following contact with copper and its salts have appeared in the literature, however the exposure concentrations leading to any effect have been poorly characterized. In studies, the possible contamination with nickel (which causes allergies definitely) has been raised as a reason for any reactions observed.

Entry into the blood-stream, via for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

There is some evidence to suggest that the material may cause mild but significant inflammation of the skin either following direct contact or after a delay of some time. Repeated exposure can cause contact dermatitis which is characterised by redness, swelling and blistering.

Skin contact with the material may be harmful; systemic effects may result following absorption.

**Eye Contact:** Copper salts, in contact with the eye, may produce inflammation of the conjunctiva, or even ulceration and cloudiness of the cornea. If applied to the eyes, this material causes severe eye damage.

### CHRONIC HEALTH EFFECTS

Principal routes of exposure include accidental contact with the molten metal and inhalation of fume arising as a consequence

of the action of the flame on the rod / wire. Although fume generation rates are generally low, excessive heating of the material, well above its quoted melting point, may result in over-exposure.

Harmful: danger of serious damage to health by prolonged exposure through inhalation.

There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population.

Copper has fairly low toxicity. Some rare hereditary conditions (Wilson disease or hepatolenticular degeneration) can lead to accumulation of copper on exposure, causing irreversible damage to a variety of organs (liver, kidney, CNS, bone, vision) and lead to death. There may be anaemia and cirrhosis of the liver.

Metallic dusts generated by the industrial process give rise to a number of potential health problems. The larger particles,

above 5 micron, are nose and throat irritants. Smaller particles however, may cause lung deterioration.

Particles of less than

1.5 micron can be trapped in the lungs and, dependent on the nature of the particle, may give rise to further serious health consequences.

### Additional toxicological information:



Organic polymers may be used in the manufacture of various welding consumables. Overexposure to their decomposition by products may result in a condition known as polymer fume fever. Polymer fume fever usually occurs within 4 to 8 hours of exposure with the presentation of flu like symptoms, including mild pulmonary irritation with or without an increase in body temperature. Signs of exposure can include an increase in white blood cell count. Resolution of symptoms typically occurs quickly, usually not lasting longer than 48 hours.

**Acute effects (acute toxicity, irritation and corrosivity):**

Harmful if swallowed.

Causes serious eye damage.

Irritating to skin.

**Repeated Dose Toxicity:** Danger of very serious irreversible effects.

**Germ cell mutagenicity** Based on available data, the classification criteria are not met.

**Carcinogenicity** Based on available data, the classification criteria are not met.

**Reproductive toxicity** Based on available data, the classification criteria are not met.

**STOT-single exposure** Based on available data, the classification criteria are not met.

**STOT-repeated exposure** Causes damage to the respiratory system through prolonged or repeated exposure.

Route of exposure: Inhalation.

**Aspiration hazard** Based on available data, the classification criteria are not met.

**MIXTURE OF CHEMICALS**

CHEMICAL	TOXICITY	IRRITATION
Brazing Alloys	Not Available	Not Available
Copper	Dermal (rat) LD50: >2000 mg/kg [1] Inhalation (rat) LC50: 0.733 mg/l/4hr [1] Inhalation (rat) LC50: 1.03 mg/l/4hr [1] Inhalation (rat) LC50: 1.67 mg/l/4hr [1] Oral (rat) LD50: 300-500 mg/kg[1]	Nil Reported
Nickel	Oral (rat) LD50: 5000 mg/kg[2]	Not Available
Zinc	Dermal (rabbit) LD50: 1130 mg/kg[2] Oral (rat) LD50: >2000 mg/kg[1]	Not Available
Tin	Dermal (rat) LD50: >2000 mg/kg[1] Oral (rat) LD50: >2000 mg/kg[1]	Not Available
Sliver	Oral (rat) LD50: >2000 mg/kg[1]	Not Available
Silicon	Dermal (rabbit) LD50: >5000 mg/kg[1] Oral (rat) LD50: 3160 mg/kg[2]	Nil Reported

**Fumes**

Nickel Fumes	Oral (rat) LD50: 5000 mg/kg[2]	Not Available
Welding Fumes	Not Available	Not Available
Copper Fumes	Dermal (rat) LD50: >2000 mg/kg [1] Inhalation (rat) LC50: 0.733 mg/l/4hr [1] Inhalation (rat) LC50: 1.03 mg/l/4hr [1] Inhalation (rat) LC50: 1.67 mg/l/4hr [1] Oral (rat) LD50: 300-500 mg/kg[1]	Not Available
Zinc Oxide Fumes	Oral (rat) LD50: >5000 mg/kg [1]	Eye (rabbit): 500 mg/24h Mild Skin (rabbit): 500 mg/24h Mild

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.\* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

## 12 Ecological information

Ingredient	96 hr LC50 Fish, mg/l	48 hr EC50 Crustacea, mg/l	ErC50 Algae, mg/l
Copper - (7440-50-8)	0.0103, Pimephales promelas	0.0025, Daphnia magna	0.018 (72 hr), Pseudokirchneriella subcapitata
Zinc – (7440-66-6)	0.182, Oncorhynchus tshawytscha	0.068, Daphnia magna	0.106 (72 hr), Pseudokirchneriella subcapitata
Nickel - (7440-02-0)	Not Available	Not Available	Not Available
Tin - (7440-31-5)	Not Available	Not Available	Not Available
Silicon - (7440-21-3)	Not Available	Not Available	Not Available
Silver	Not Available	Not Available	Not Available
Welding Fumes	Not Available	Not Available	Not Available
Copper Fumes	Not Available	Not Available	Not Available
Zinc Oxide Fumes	Not Available	Not Available	Not Available
Nickel Fumes	Not Available	Not Available	Not Available

### Copper, Zinc, Nickel, Tin, Silicon

#### Toxicity

Very toxic to aquatic life.

Toxic to aquatic life with long lasting effects.

No additional information provided for this product.

#### Persistence and degradability

There is no data available on the preparation itself.

#### Bioaccumulative potential

Not Measured

#### Mobility in soil

No data available.

**PBT and vPvB assessment** This product contains no PBT/vPvB chemicals.

#### Other adverse effects

No data available.

## 13 Disposal considerations

### Waste treatment methods

#### Recommendation:

The user of this material has the responsibility to dispose of unused material, residues and containers in compliance with all relevant local, state and federal laws and regulations regarding treatment, storage and disposal for hazardous and nonhazardous wastes.

#### Uncleaned packagings:

**Recommendation:** Disposal must be made according to official regulations.

## 14 Transport Information

HAZCHEM	Not Applicable
UN-Number DOT, ADR, ADN, IMDG, IATA	Not Regulated
UN proper shipping name	Not Regulated

DOT, ADR, ADN, IMDG, IATA	
Transport hazard class(es) DOT, ADR, ADN, IMDG, IATA Class	Not Regulated
Packing group DOT, ADR, IMDG, IATA	Not Regulated
Environmental hazards: Marine pollutant:	Yes
Special precautions for user	Not applicable.
Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code	Not applicable.
UN "Model Regulation":	Not regulated.

## 15 Regulatory information

### Product Name: Silver Brazing Alloys

Refer to the Australian Inventory of Chemical Substances – AICS at <https://www.nicnas.gov.au/chemicals-on-AICS#main>

**Poison schedule:** Classified as a Schedule 6 (S6) Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP). <https://www.legislation.gov.au/Details/F2016L01638>

**Classifications:** Safework Australia criteria is based on the Globally Harmonised System (GHS) of Classification and Labelling of Chemicals.

The classifications and phrases listed below are based on the Approved Criteria for Classifying Hazardous Substances [NOHSC: 1008(2004)].

## 16 Other information

### References

Preparation of Safety Data Sheets for Hazardous Chemicals Code of Practice

Standard for the Uniform Scheduling of Medicines and Poisons

Australian Code for the Transport of Dangerous Goods by Road & Rail.

Modell Work Health and Safety Regulations, Schedule 10: Prohibited carcinogens, restricted carcinogens and restricted hazardous chemicals.

Workplace exposure standards for airborne contaminants, Safe work, Australia

American Conference of Industrial Hygienists (ACGIH)

Globally Harmonised System of classification and labelling of chemicals.

**WELDING (1):** Due to the diversity of welding techniques, processes, materials used, nature of the surface being welded and the presence of contaminants, the fumes & gases associated with welding will vary in composition and quantity. When assessing a welding process, the toxic fumes generated may not only be associated with the parent metal, filler wire or electrode. The welding/cutting arc may generate nitrogen oxides, carbon monoxide & other gases, whilst UV radiation emitted from some arcs generates ozone. Ozone may irritate mucous membranes and cause pulmonary oedema & haemorrhage. Shielding gases (e.g. carbon

dioxide and inert gases i.e. argon and helium) in high concentrations, in confined spaces, may reduce oxygen in the atmosphere to dangerous levels, resulting in possible asphyxiation.

**WELDING (2):** In addition to complying with individual exposure standards for specific contaminants, where current manual welding processes are used, the fume concentration inside the welder's helmet should not exceed 5 mg/m<sup>3</sup> ( unless otherwise classified) when collected in accordance with Australian Standard AS 3853.1: Fume from welding and allied processes - Guide to methods for the sampling and analysis of particulate matter and AS 3853.2: Fume from welding and allied processes - Guide to methods for the sampling and analysis of gases. Airway irritation and metal fume fever are the most common acute effects from welding fumes. Reported to cause reduced sperm quality in welders.

**WELDING (3):** Other gases and fumes associated with welding processes include: Inert shielding gases (e.g. argon, carbon dioxide, helium) which may reduce the atmospheric oxygen content in poorly ventilated areas. UV-radiation and Infra-Red radiation may decompose chlorinated degreasing agents to form highly toxic and irritating phosgene gas. This may occur if a metal has been degreased but inadequately dried or when vapours from a nearby degreasing bath enter the welding zone.

**WELDING (4):** Welding fumes may contain a wide variety of chemical contaminants, including oxides and salts of metals and other compounds which may be generated from electrodes, filler wire, flux materials and from the welded material (e.g. painted surfaces). Welding stainless-steel and its alloys generates nickel and chromium (VI) compounds. Welding fumes are retained in the lungs. Sparingly soluble compounds may be released slowly from the lungs. Welding fume is classified as possibly carcinogenic to humans (IARC Group 2B).

#### **PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:**

The recommendation for protective equipment contained within this report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

#### **Disclaimer:**

We urge each end user and recipient of this SDS to study it carefully. If necessary, consult an industrial hygienist or other expert to understand this information and safeguard the environment and protect workers from potential hazards associated with the handling or use of this product.

Harris Products Group cannot anticipate all conditions under which this information and its product, or the products of other manufacturers in combination with its product, may be used. It is the user's responsibility to ensure safe conditions for use, handling, storage and disposal of the product, and to assume liability for loss, injury, damage or expense due to improper use. The information in the sheet was written based on the best knowledge and experience currently available.

#### **WARNING: PRODUCT COMPONENTS PRESENT HEALTH AND SAFETY HAZARDS. READ AND UNDERSTAND THIS SAFETY DATA SHEET (S.DS.). ALSO, FOLLOW YOUR EMPLOYER'S SAFETY PRACTICES.**

The information contained herein relates only to the specific product. If the product is combined with other materials, all component properties must be considered. **BE SURE TO CONSULT THE LATEST VERSION OF THE SDS. SAFETY DATA SHEETS ARE AVAILABLE FROM HARRIS PRODUCTS GROUP** Harris Products Group, HGE PTY LTD, Brisbane | Melbourne | Perth | New Zealand, 14 Queensland Rd, Darra, QLD 4076, Phone: (07) 3375 3670 | Fax: (07) 3375 3620, Email: sales@hgea.com.au, www.harrisproductsgroup.com.au,

#### **STATEMENT OF LIABILITY-DISCLAIMER**

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